

Questions with Answer Keys

MathonGo

Q1: 24 Feb (Shift 1) - Single Correct

Given below are two statements :

Statement-I: Two photons having equal linear momenta have equal wavelengths.

Statement-II: If the wavelength of photon is decreased, then the momentum and energy of a photon will also decrease.

In the light of the above statements, choose the correct answer from the options given below.

- (1) Statement-I is false but Statement-II is true
- (2) Both Statement-I and Statement-II are true
- (3) Both Statement-I and Statement-II are false
- (4) Statement-I is true but Statement-II is false

Q2: 24 Feb (Shift 1) - Numerical

A proton and a Li^{3+} nucleus are accelerated by the same potential. If λ_{Li} and λ_{p} denote the de Broglie wavelengths of Li^{3+} and proton respectively, then the value of $\frac{\lambda_{\text{Li}}}{\lambda_{\text{p}}}$ is $x \times 10^{-1}$. The value of x is _____.

[Rounded off to the nearest integer] [Mass of $\text{Li}^{3+} = 8.3$ mass of proton]

Q3: 24 Feb (Shift 2) - Single Correct

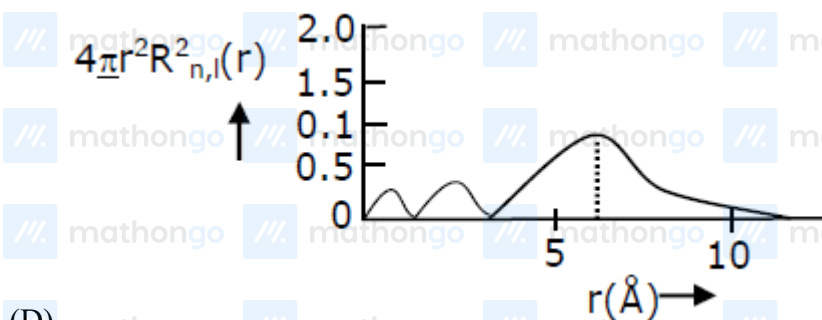
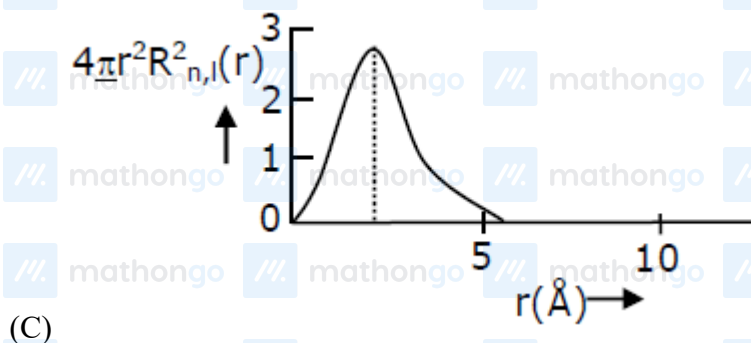
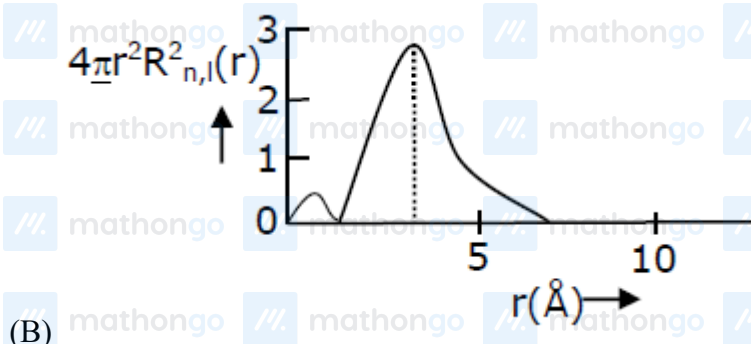
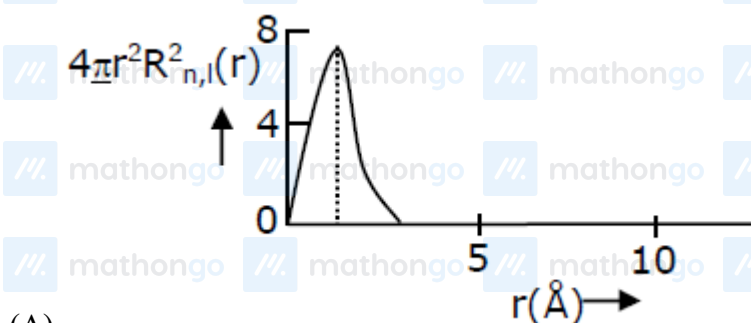
According to Bohr's atomic theory :

- (A) Kinetic energy of electron is $\propto \frac{Z^2}{n^2}$
- (B) The product of velocity (v) of electron and principal quantum number (n). ' vn ' $\propto Z^2$.
- (C) Frequency of revolution of electron in an orbit is $\propto \frac{Z^3}{n^3}$.
- (D) Coulombic force of attraction on the electron is $\propto \frac{Z^3}{n^4}$. Choose the most appropriate answer from the options given below:

- (1) (C) only
- (2) (A) and (D) only
- (3) (A) only
- (4) (A), (C) and (D) only

Q4: 25 Feb (Shift 1) - Single Correct

The plots of radial distribution functions for various orbitals of hydrogen atom against 'r' are given below:



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The correct plot for 3 s orbital is:

- (1) D
- (2) B
- (3) A
- (4) C

Q5: 25 Feb (Shift 2) - Numerical

Electromagnetic radiation of wavelength 663nm is just sufficient to ionize the atom of metal A. The ionization energy of metal A in kJmol^{-1} is _____. (Rounded off to the nearest integer)

$$\left[h = 6.63 \times 10^{-34} \text{ Js}, c = 3.00 \times 10^8 \text{ ms}^{-1}, N_A = 6.02 \times 10^{23} \text{ mol}^{-1} \right]$$

Q6: 26 Feb (Shift 1) - Single Correct

The orbital having two radial as well as two angular nodes is

- (1) 5 d
- (2) 4f
- (3) 3p
- (4) 4 d

Q7: 26 Feb (Shift 2) - Numerical

A ball weighing 10 g is moving with a velocity of 90 ms^{-1} . If the uncertainty in its velocity is 5%, then the uncertainty in its position is _____ $\times 10^{-33}$ m. (Rounded off to the nearest integer) [Given :

$$h = 6.63 \times 10^{-34} \text{ Js}]$$

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Answer Key

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Q1 (4)

Q2 (2)

Q3 (2)

Q4 (1)

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Q5 (180)

Q6 (1)

Q7 (1)

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