

Q1 - 25 July - Shift 2

For micelle formation, which of the following statements are correct?

(A) Micelle formation is an exothermic process.

(B) Micelle formation is an endothermic process.

(C) The entropy change is positive.

(D) The entropy change is negative.

(A) A and D only (B) A and C only

(C) B and C only (D) B and D only

Space for your notes:

Q2 - 26 July - Shift 2

Questions

MathonGo

Given below are two statements : one is labelled as Assertion A and the other is labelled as Reason R.

Assertion A : Finest gold is red in colour, as the size of the particles increases, it appears purple then blue and finally gold.

Assertion R : The colour of the colloidal solution depends on the wavelength of light scattered by the dispersed particles.

In the light of the above statements, choose the most appropriate answer from the options given

below;

(A) Both A and R are true and R is the correct explanation of A

(B) Both A and R are true but R is NOT the correct explanation of A

(C) A is true but R is false

(D) A is false but R is true

Space for your notes:

Q3 - 27 July - Shift 2

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Questions

MathonGo

Given below are two statements: one is labelled as **Assertion (A)** and the other is labelled as **Reason (R)**

Space for your notes:

Assertion (A) : Dissolved substances can be removed from a colloidal solution by diffusion through a parchment paper.

Reason (R) : Particles in a true solution cannot pass through parchment paper but the colloidal particles can pass through the parchment paper.

In the light of the above statements, choose the correct answer from the options given below:

- (A) Both (A) and (R) are correct and (R) is the correct explanation of (A)
- (B) Both (A) and (R) are correct but (R) is not the correct explanation of (A)
- (C) (A) is correct but (R) is not correct
- (D) (A) is not correct but (R) is correct

Q4 - 28 July - Shift 1

Statements about Enzyme Inhibitor Drugs are given below :

Space for your notes:

- (A) There are Competitive and Non-competitive inhibitor drugs.
- (B) These can bind at the active sites and allosteric sites.
- (C) Competitive Drugs are allosteric site blocking drugs.
- (D) Non-competitive Drugs are active site blocking drugs.

Choose the correct answer from the options given below :

- (A) (A), (D) only
- (B) (A), (C) only
- (C) (A), (B) only
- (D) (A), (B), (C) only

Q5 - 28 July - Shift 2

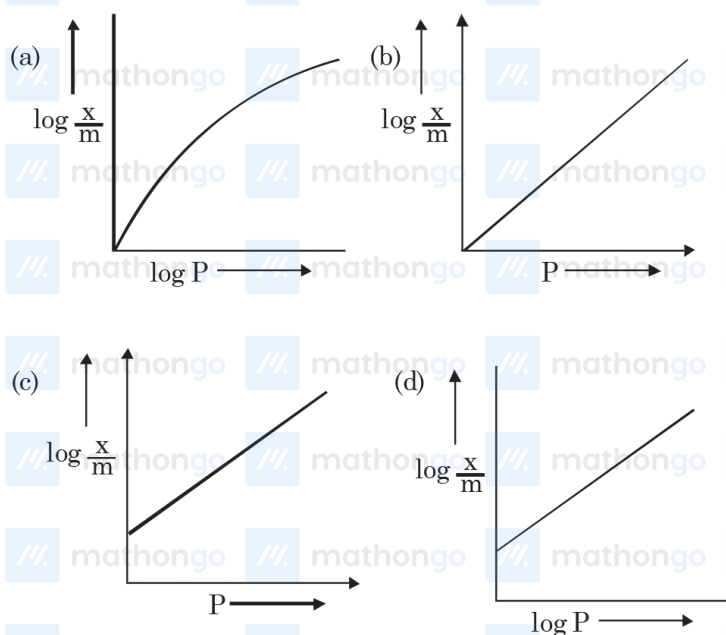
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Questions

MathonGo

Among the following the number of curves not in accordance with Freundlich adsorption isotherm is

Space for your notes:



Q6 - 29 July - Shift 1

100 mL of 5% (w/v) solution of NaCl in water was prepared in 250 mL beaker. Albumin from the egg was poured into NaCl solution and stirred well.

Space for your notes:

This resulted in a/an :

- (A) Lyophilic sol (B) Lyophobic sol
(C) Emulsion (D) Precipitate

Q7 - 29 July - Shift 2

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Questions

MathonGo

Given below are the critical temperatures of some of the gases :

Space for your notes:

Gas	Critical temperature (K)
He	5.2
CH ₄	190
CO ₂	304.2
NH ₃	405.5

The gas showing least adsorption on a definite amount of charcoal is :

- (A) He (B) CH₄
(C) CO₂ (D) NH₃

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Questions

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Answer Key

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Q1 (C)**Q2 (A)****Q3 (C)****Q4 (C)**

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Q5 (3)**Q6 (A)****Q7 (A)**

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Q1 (C)

For micelle formation, $\Delta S > 0$ (hydrophobic effect) This is possible because, the decrease in entropy due to clustering is offset by increase in entropy due to desolvation of the surfactant, Also $\Delta H > 0$

Q2 (A)**(A)****Q3 (C)**

Assertion (A): Correct.

Reason(R): Incorrect.

Particles of true solution pass through parchment paper thus answer is (C).

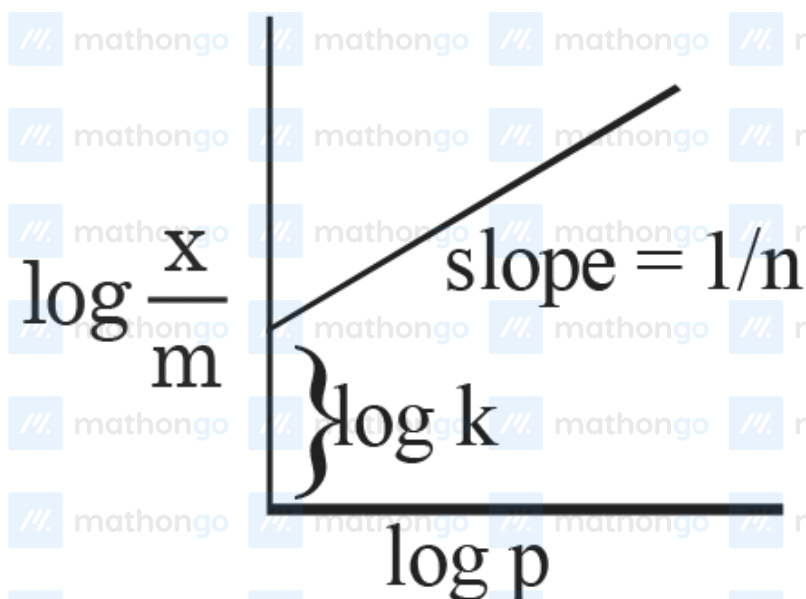
Q4 (C)

Enzyme inhibitors can be competitive inhibitors (inhibit the attachment of substrate on active site of enzyme) and non-competitive inhibitor (changes the active site of enzyme after binding at allosteric site.)

Q5 (3)

$$\frac{x}{m} = KP^n$$

$$\log \frac{x}{m} = \frac{1}{n} \log p + \log k$$



Q6 (A)

Standard method for the preparation of lyophilic sol. (Discussed in lab Manual)

Q7 (A)

More the critical temp. of gas greater is the ease of liquefaction hence greater is the adsorption.