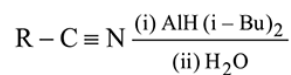


JEE Mains 2019 Chapter wise Question Bank

Aldehydes and Ketones - Questions

Q1

The major product of following reaction is:

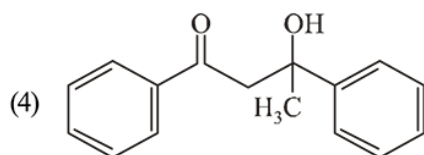
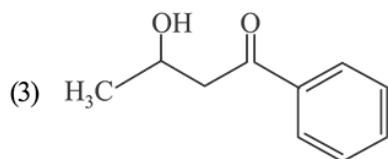
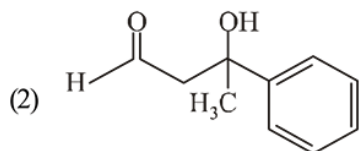
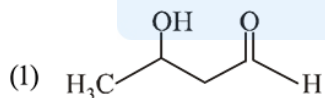
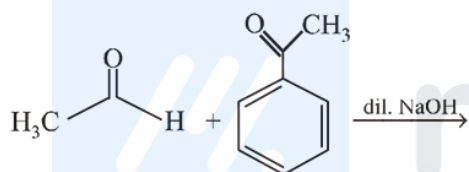


- (1) RCOOH (2) RCONH₂
 (3) RCHO (4) RCH₂NH₂

9 Jan Morning

Q2

The major product formed in the following reaction is:



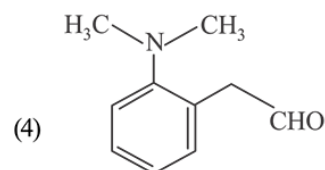
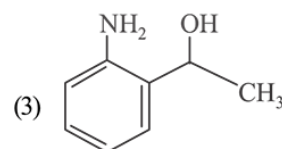
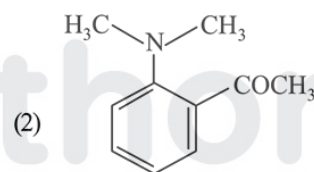
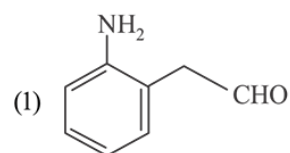
9 Jan Evening

Q3

The tests performed on compound X and their inferences are:

Test	Inference
(a) 2, 4 - DNP test	Coloured precipitate
(b) Iodoform test	Yellow precipitate
(c) Azo-dye test	No dye formation

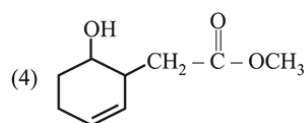
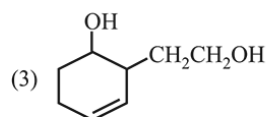
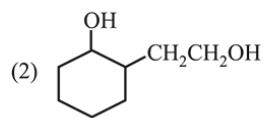
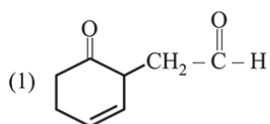
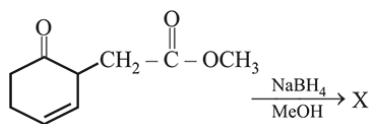
Compound 'X' is:



9 Jan Evening

Q4

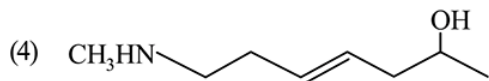
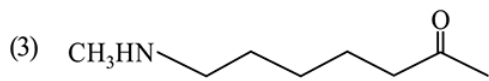
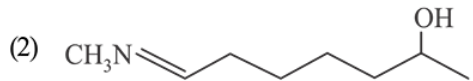
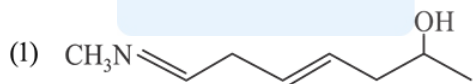
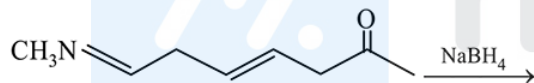
Aldehydes and Ketones



10 Jan Morning

Q5

The major product of the following reaction is:

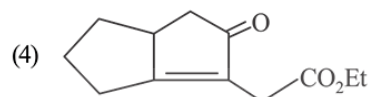
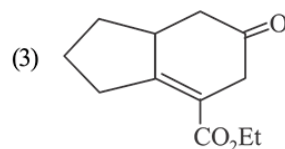
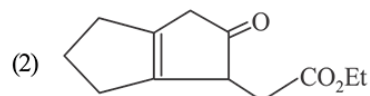
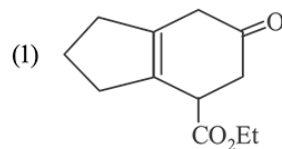
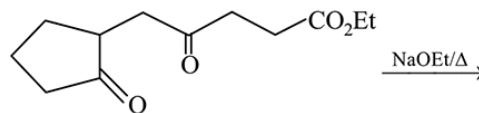


10 Jan Evening

Q6

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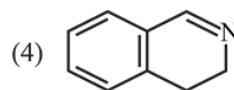
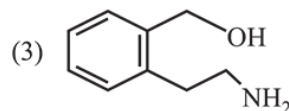
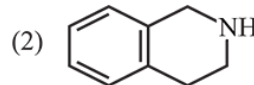
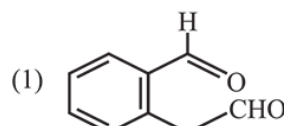
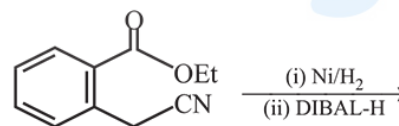
The major product obtained in the following reaction is:



10 Jan Evening

Q7

The major product of the following reaction is :

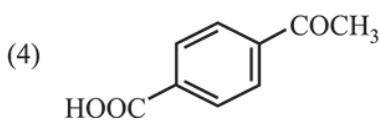
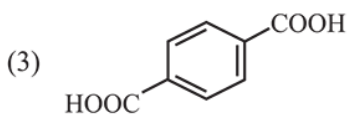
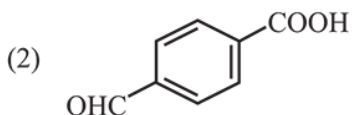
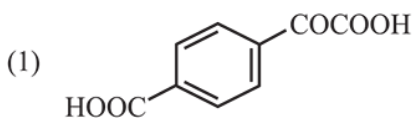
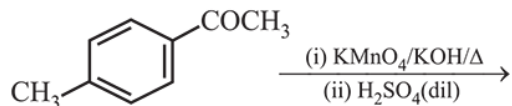


11 Jan Morning

Q8

Aldehydes and Ketones

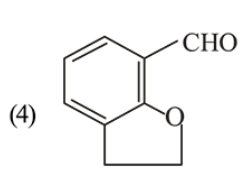
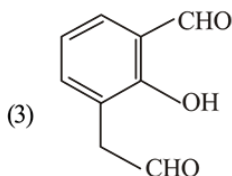
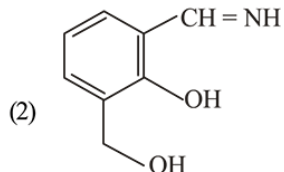
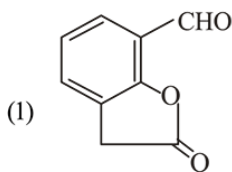
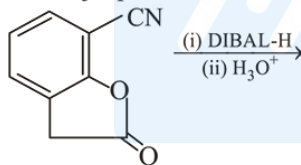
The major product of the following reaction is :



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Q9

The major product of the following reaction is:



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Q10

JEE Mains 2019 Chapter wise Question Bank

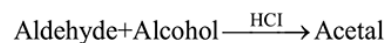
$\text{CH}_3\text{CH}_2\text{-}\overset{\text{OH}}{\underset{\text{Ph}}{\text{C}}}\text{-CH}_3$ cannot be prepared by:

- (1) $\text{CH}_3\text{CH}_2\text{COCH}_3 + \text{PhMgX}$
- (2) $\text{PhCOCH}_2\text{CH}_3 + \text{CH}_3\text{MgX}$
- (3) $\text{PhCOCH}_3 + \text{CH}_3\text{CH}_2\text{MgX}$
- (4) $\text{HCHO} + \text{PhCH}(\text{CH}_3)\text{CH}_2\text{MgX}$

12 Jan Morning

Q11

In the following reaction



Aldehyde	Alcohol
HCHO	^t BuOH
CH ₃ CHO	MeOH

The best combination is:

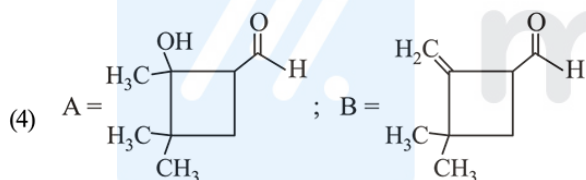
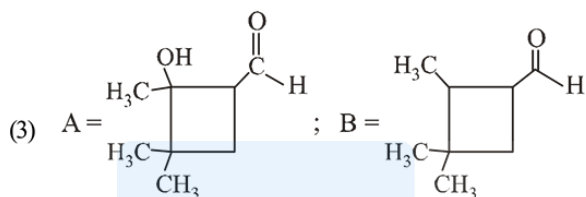
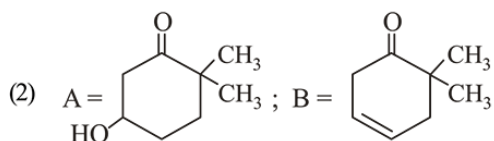
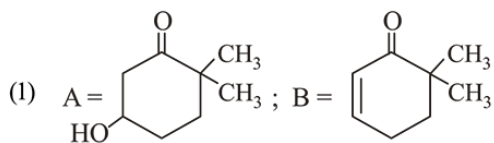
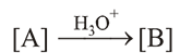
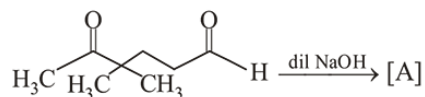
- (1) CH₃CHO and ^tBuOH
- (2) HCHO and MeOH
- (3) CH₃CHO and MeOH
- (4) HCHO and ^tBuOH

12 Jan Morning

Q12

Aldehydes and Ketones

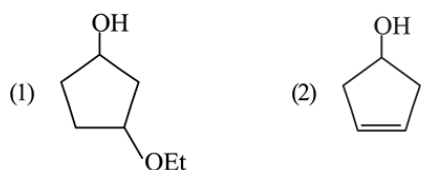
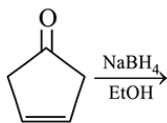
In the following reactions, products A and B are:



12 Jan Morning

Q13

The major product of the following reaction is :

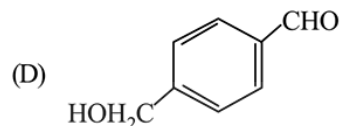
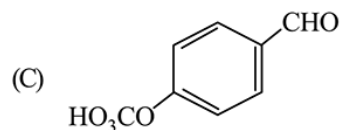
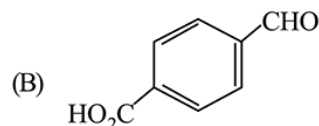
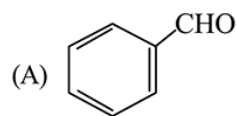


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Q14

The aldehydes which will **not** form Grignard product with one equivalent Grignard reagents are :



(1) (B), (D)

(2) (B), (C)

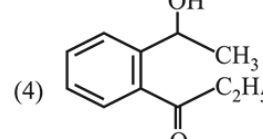
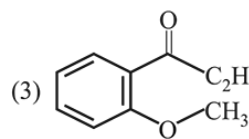
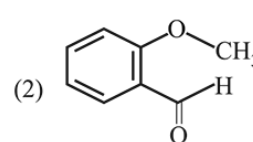
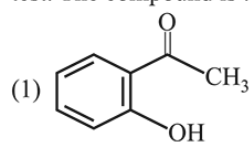
(3) (B), (C), (D)

(4) (C), (D)

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Q15

An organic compound neither reacts with neutral ferric chloride solution nor with Fehling solution. It however, reacts with Grignard reagent and gives positive iodoform test. The compound is :

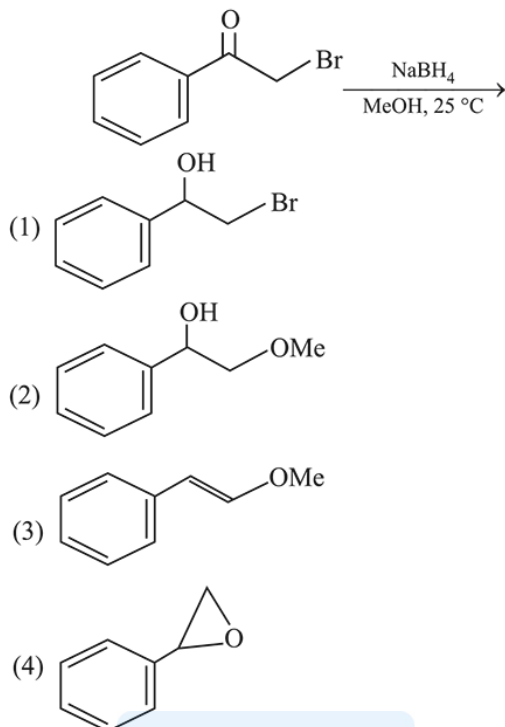


8 April Morning

Q16

Aldehydes and Ketones

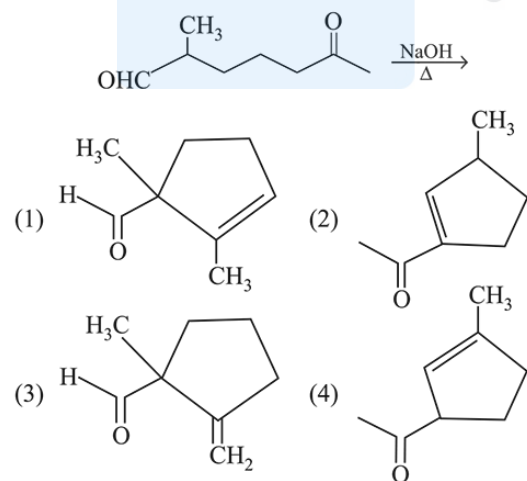
The major product of the following reaction is :



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Q17

The major product obtained in the following reaction is :

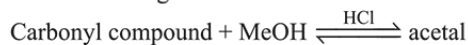


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Q18

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In the following reaction



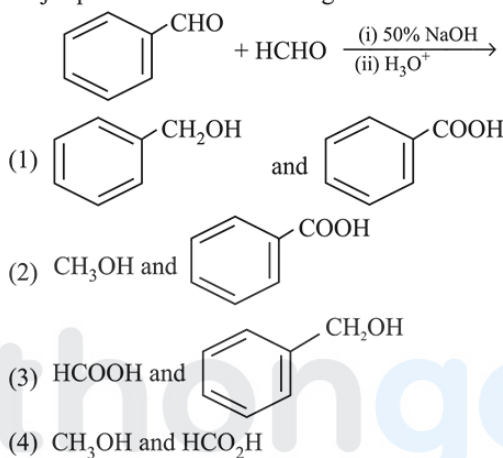
Rate of the reaction is the highest for:

- (1) Acetone as substrate and methanol in excess.
- (2) Propanal as substrate and methanol in stoichiometric amount.
- (3) Propanal as substrate and methanol in excess.
- (4) Acetone as substrate and methanol in stoichiometric amount.

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Q19

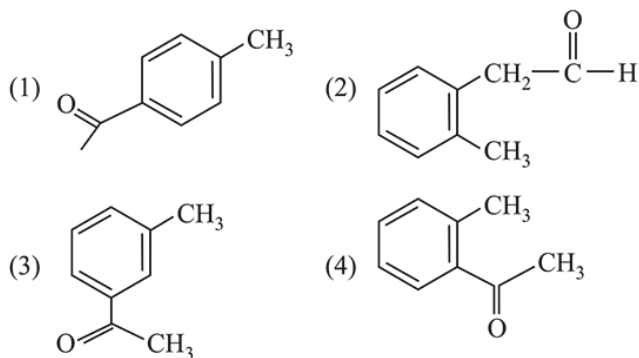
Major products of the following reaction are :



10 April Morning

Q20

Compound A ($\text{C}_9\text{H}_{10}\text{O}$) shows positive iodoform test. Oxidation of A with KMnO_4/KOH gives acid B ($\text{C}_8\text{H}_6\text{O}_4$). Anhydride of B is used for the preparation of phenolphthalein. Compound A is :

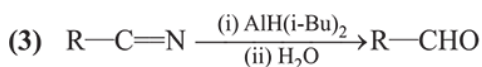


10 April Evening

JEE Mains 2019 Chapter wise Question Bank

Aldehydes and Ketones - Answers

Q1

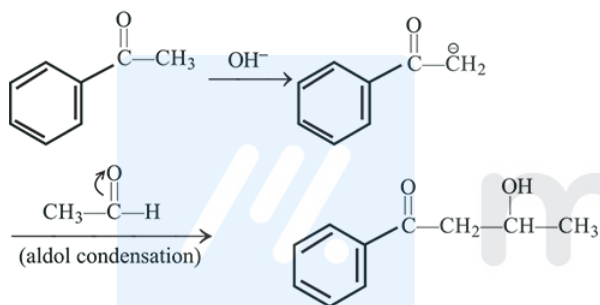


The reduction of nitriles to aldehydes can be done using DIBAL-H[AlH(i-Bu)₂].

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Q2

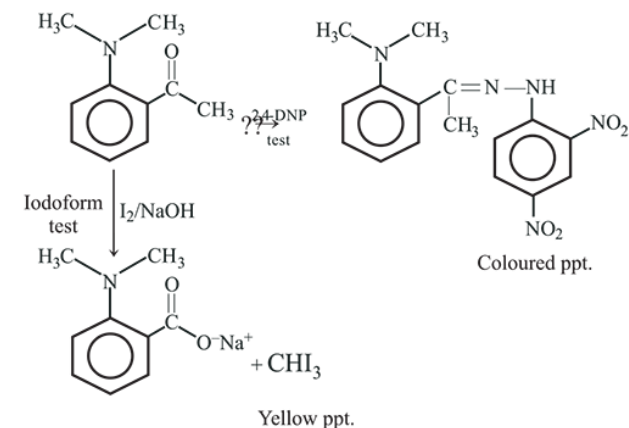
(3) Reaction mechanism involved:



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Q3

(2) Reaction involved in the tests performed is as follows:



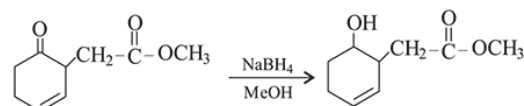
Yellow ppt.

Thus, will not give positive azo-dye test due to the presence of an electron-withdrawing group

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Q4

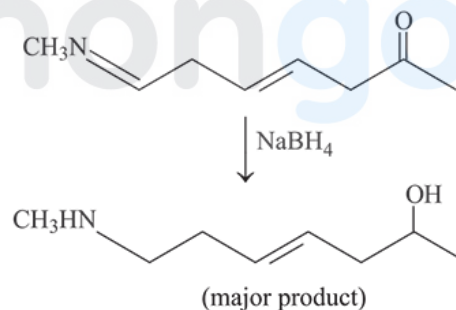
(4) NaBH₄ is a selective reducing agent, used for the reduction of aldehydes and ketones, it does not affect alkene and ester.



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Q5

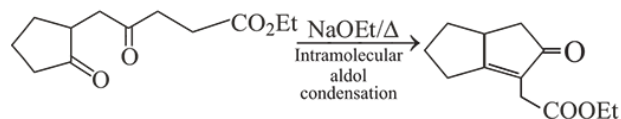
(4) Reaction involved:



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Q6

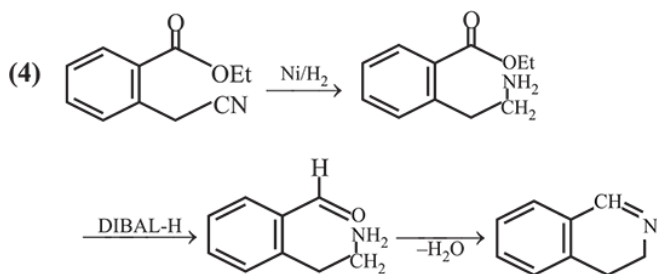
(4) Reaction involved:



10 Jan Evening

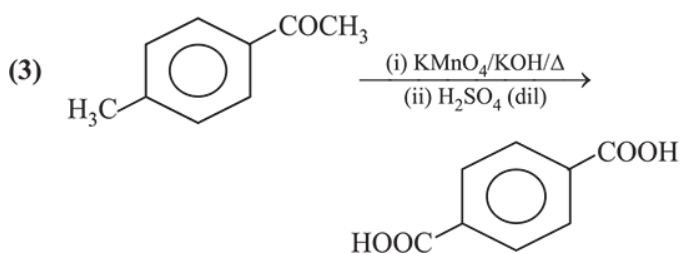
Q7

Aldehydes and Ketones



11 Jan Morning

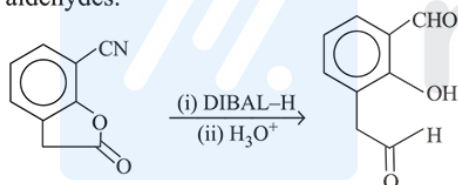
Q8



11 Jan Morning

Q9

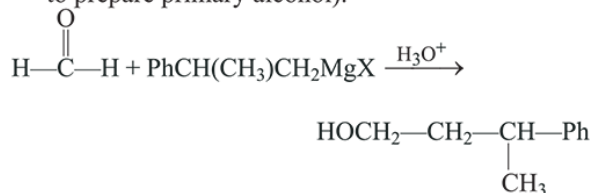
- (3) DIBAL-H will reduce cyanides and esters to aldehydes.



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Q10

- (4) Tertiary alcohol is prepared by the reaction of Grignard reagent with a ketone (formaldehyde is used to prepare primary alcohol).

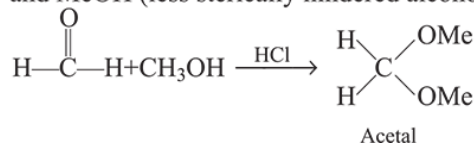


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Q11

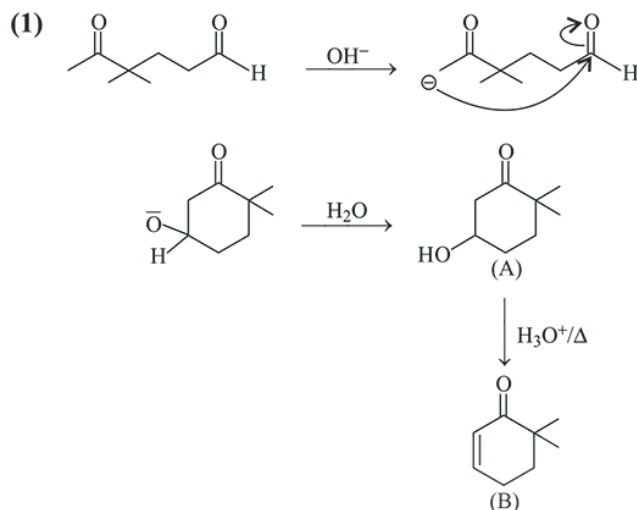
JEE Mains 2019 Chapter wise Question Bank

- (2) Best combination is HCHO (more reactive aldehyde) and MeOH (less sterically hindered alcohol).



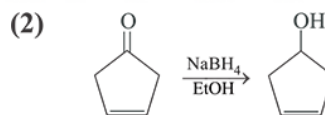
12 Jan Morning

Q12



12 Jan Morning

Q13

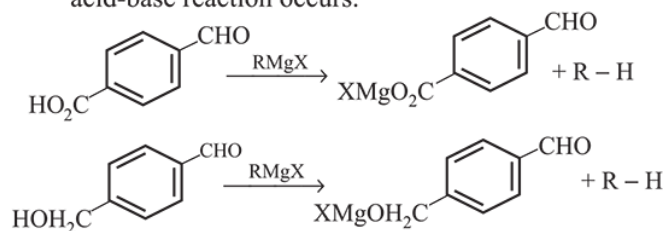


NaBH₄ does not reduce the double bond but can reduce keto group ($\text{C}=\text{O}$) into $-\text{OH}$ group.

12 Jan Evening

Q14

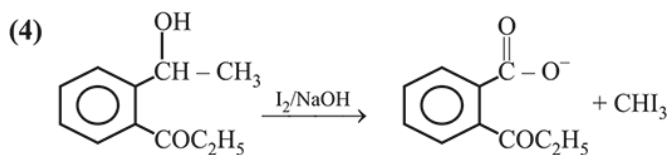
- (1) Grignard reagent will not react with aldehydes if it has a functional group which contains acidic hydrogen. Thus options (B) and (D) have $-\text{COOH}$ and $-\text{CH}_2\text{OH}$ respectively which contain acidic H-atom. Therefore, acid-base reaction occurs.



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Aldehydes and Ketones

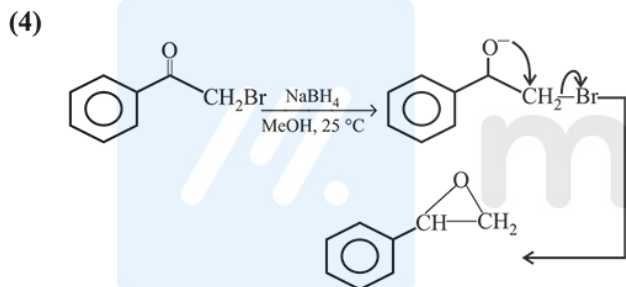
Q15



Test	Observation	Reason
Reaction with Grignard reagent	+ve	Electrophilic centre or acidic hydrogen is present
Fehling solution test	-ve	CHO group is absent
Neutral FeCl ₃ test	-ve	phenolic group is absent
Iodoform test	+ve	-COCH ₃ or -CH(OH)-CH ₃ is present

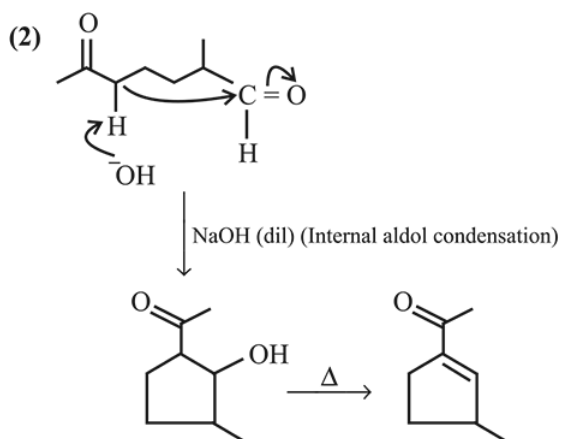
8 April Morning

Q16



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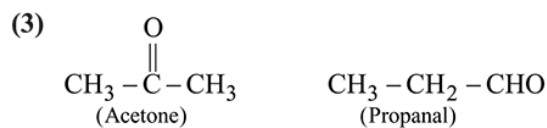
Q17



8 April Evening

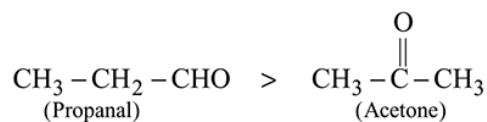
Q18

JEE Mains 2019 Chapter wise Question Bank



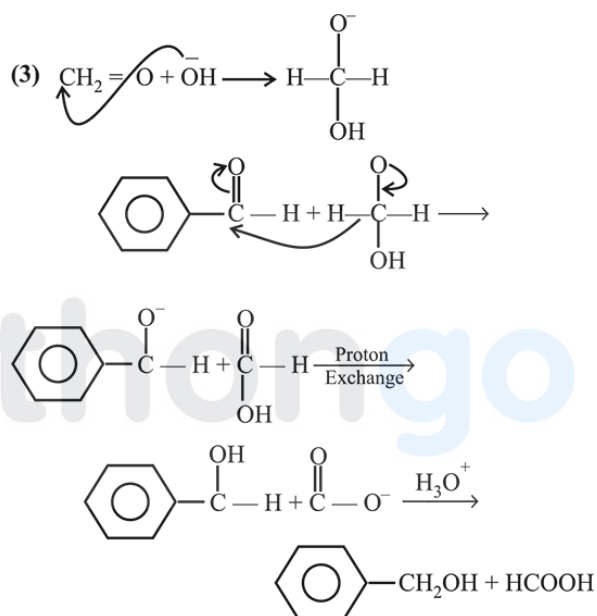
Generally, aldehydes are more reactive than ketones in nucleophilic addition reactions.

∴ Rate of reaction with alcohol to form acetal and ketal is



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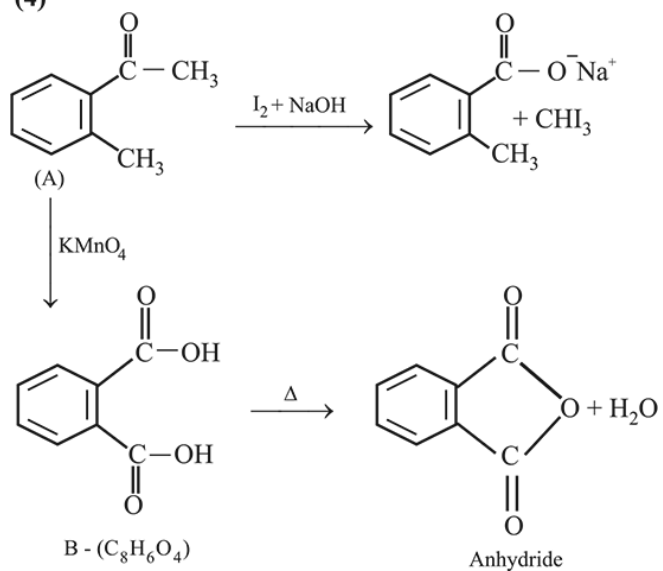
Q19



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Q20

(4)



10 April Evening



mathongo